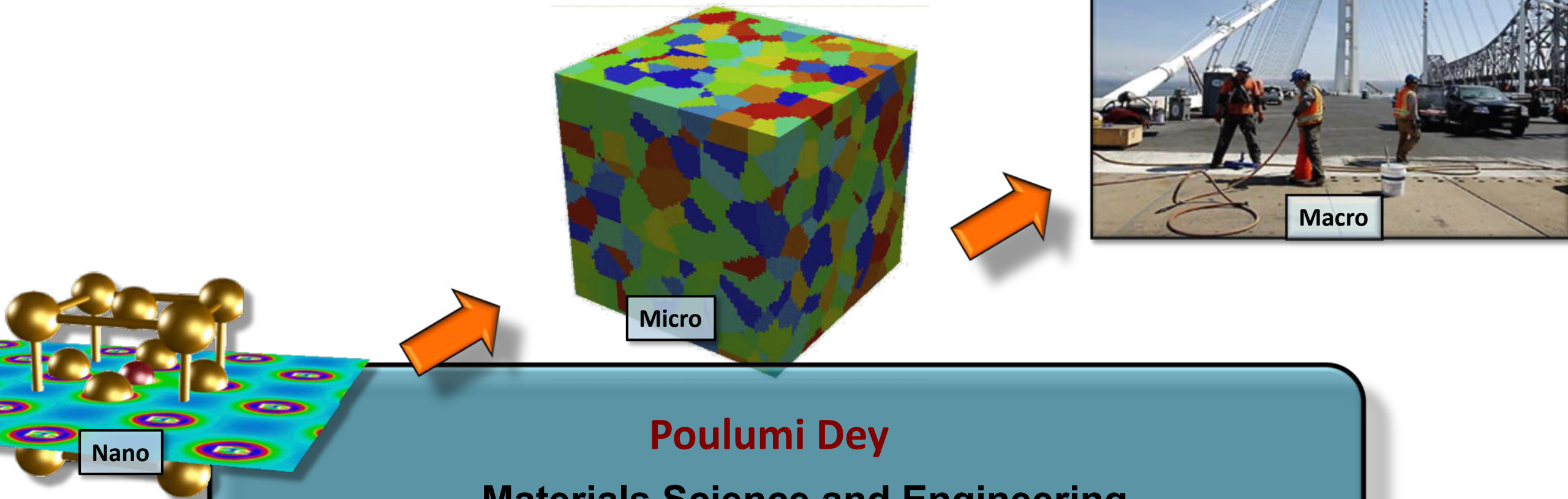


Computation-guided materials designing for sustainable & renewable energy applications



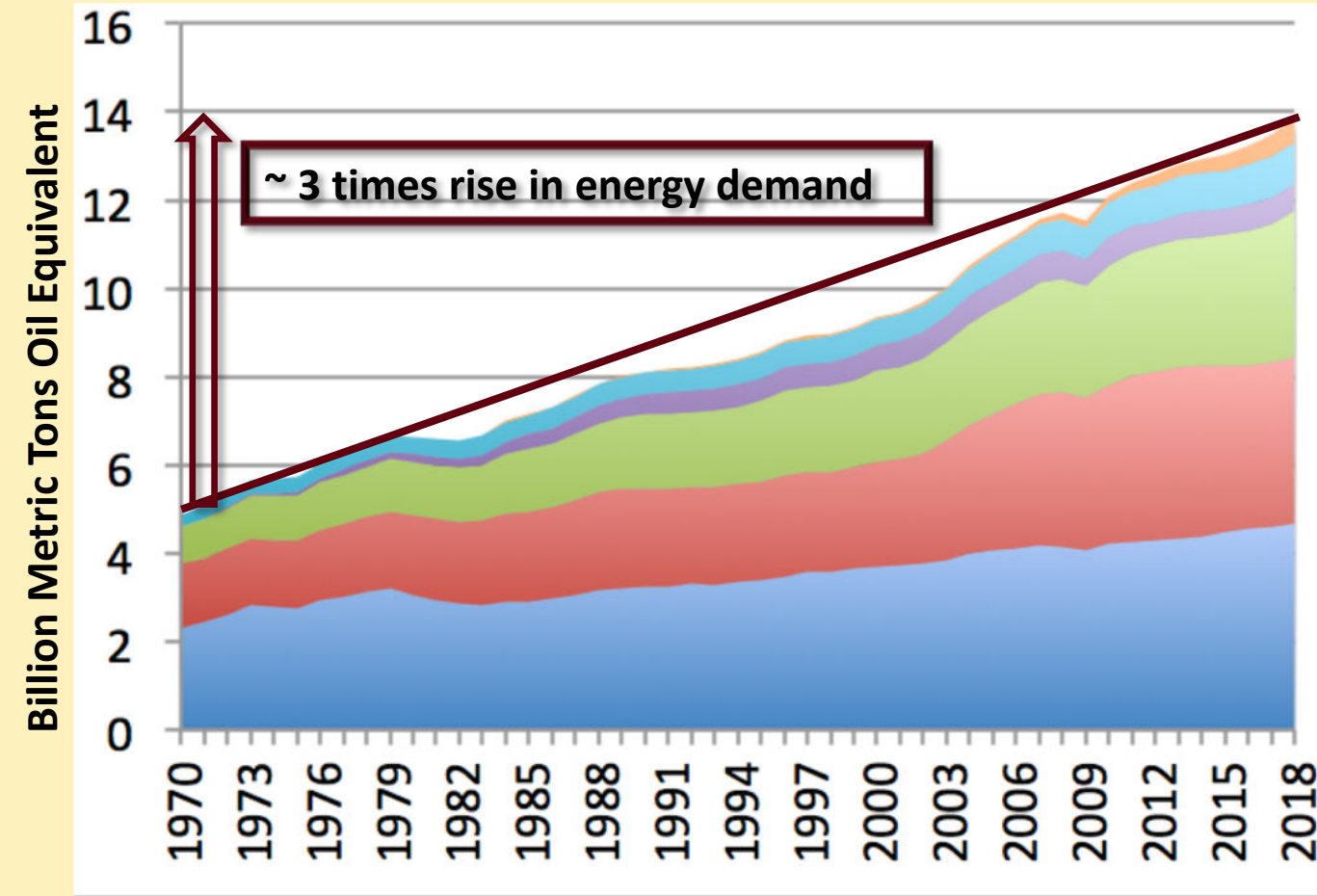
Poulumi Dey

Materials Science and Engineering

TU Delft

World Energy Consumption

(BP Statistical Review of World Energy)



What is the need of the hour?

Transition to sustainable sources of energy

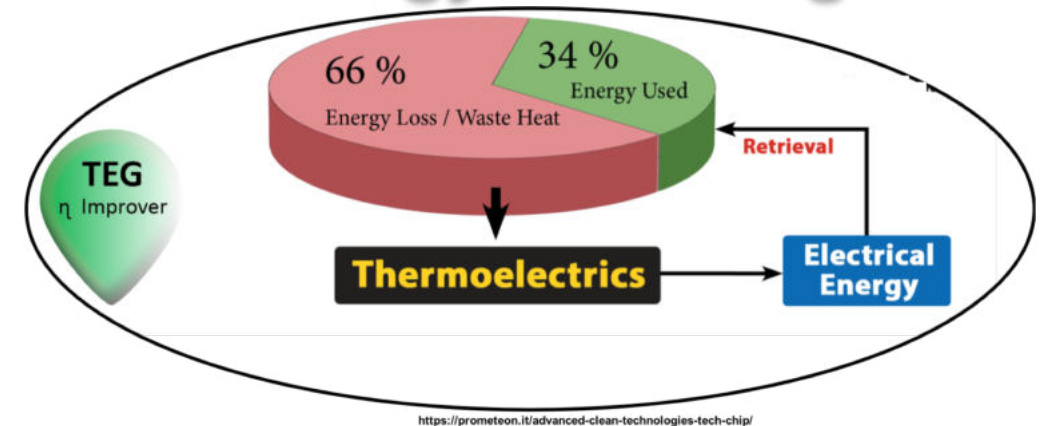
Hydrogen Energy



Storage & Transport

Research challenges?

Thermoelectricity – energy & cooling



Heat to energy conversion

HYDROGEN EFFECTS

Hydrogen Embrittlement (HE)....

West Berlin Congress Hall, 1980



Hydrogen intake causing collapse of roof

San Francisco-Oakland Bay Bridge, 2013



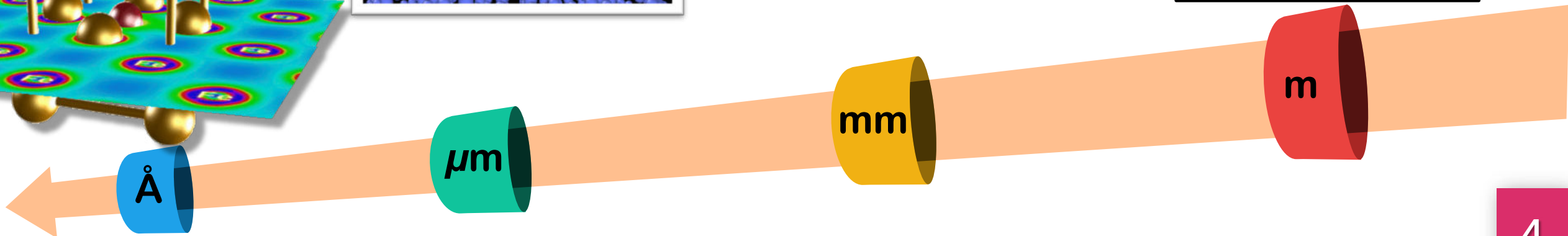
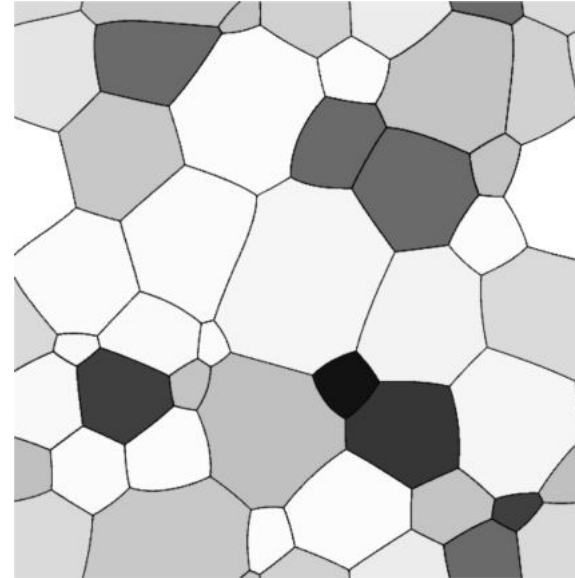
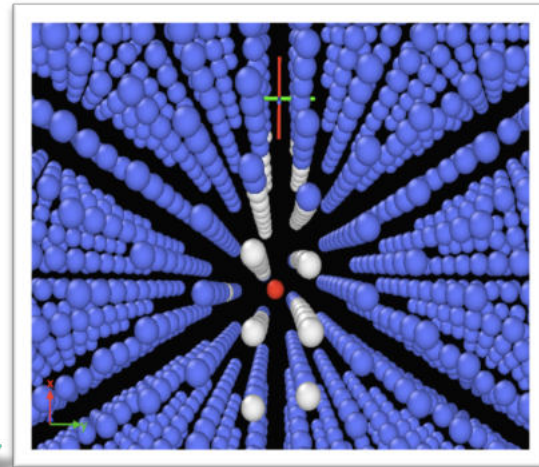
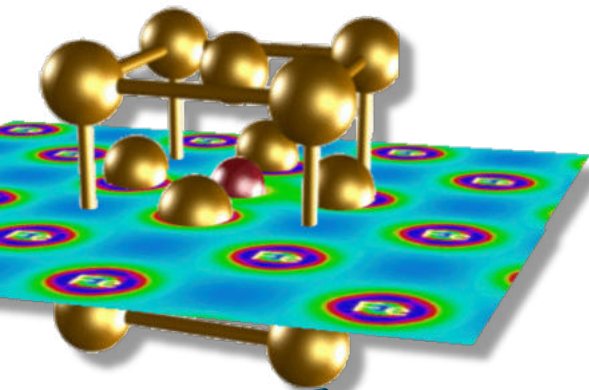
Brittle bolts and cracking due to hydrogen contamination

....big threat!

WHAT IS NEEDED?



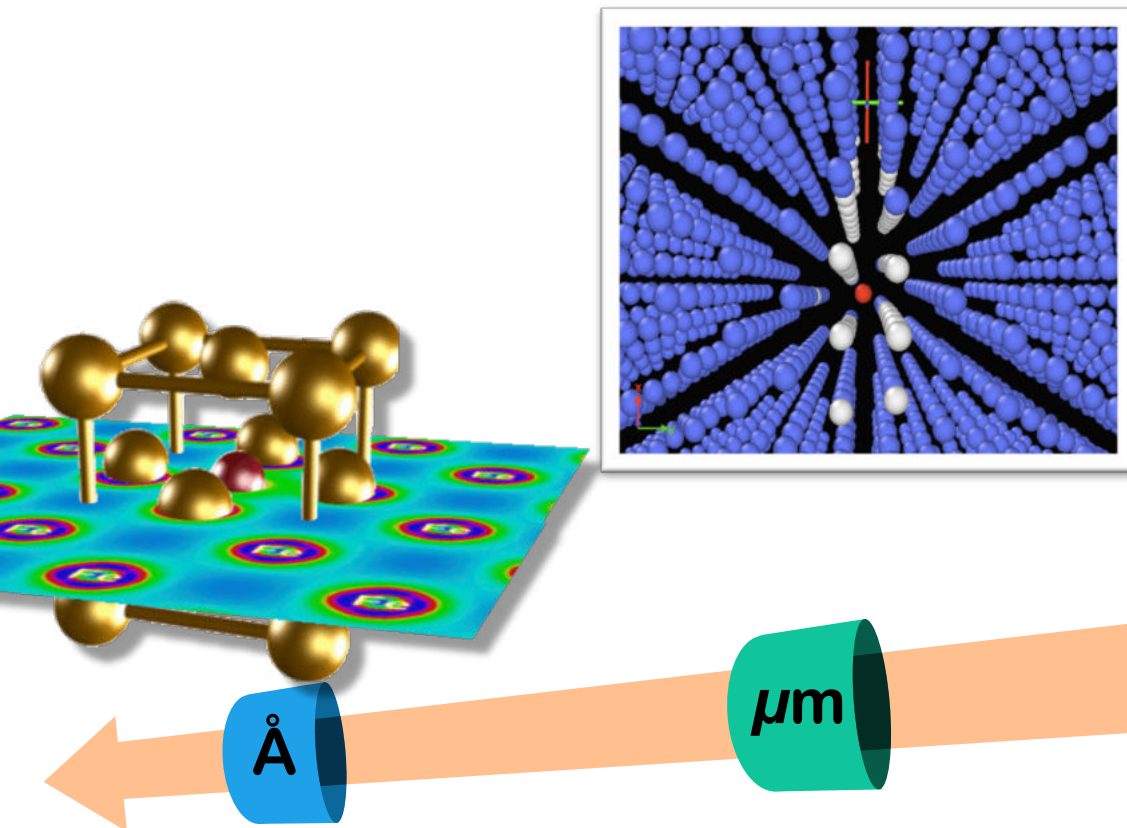
Hydrogen induced degradation



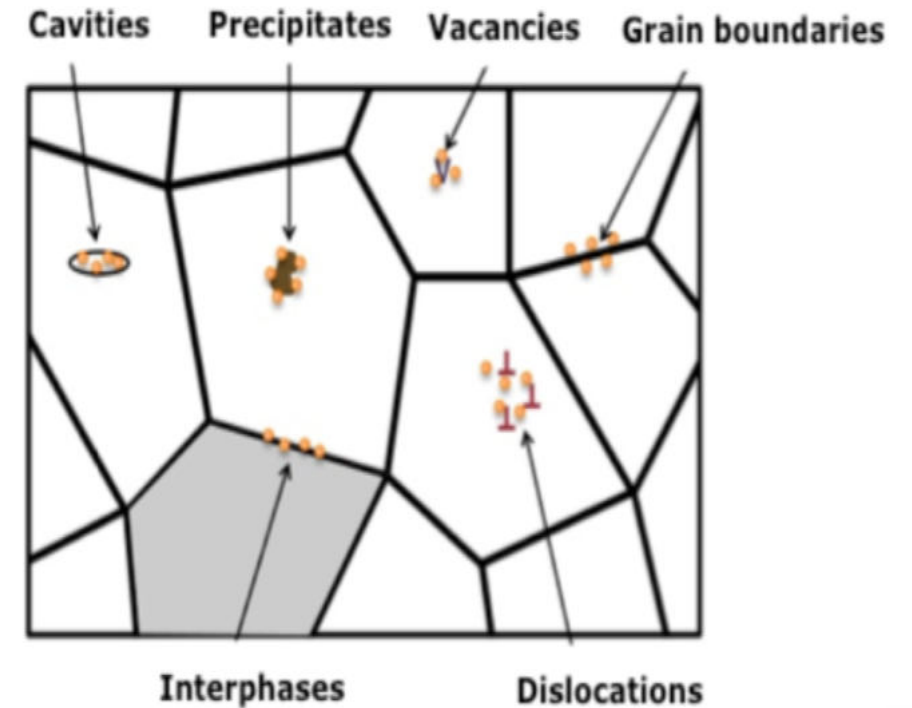
WHAT IS NEEDED?



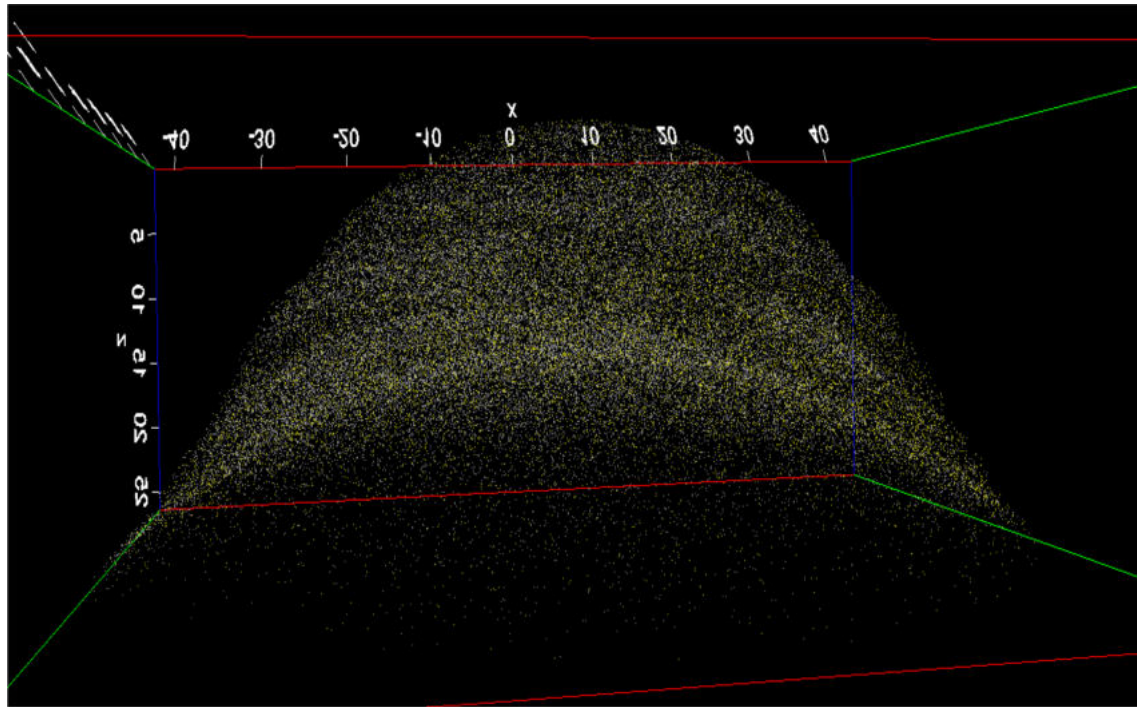
Fundamental atomistic insights
beyond experimental resolution



**H-resistant
design**

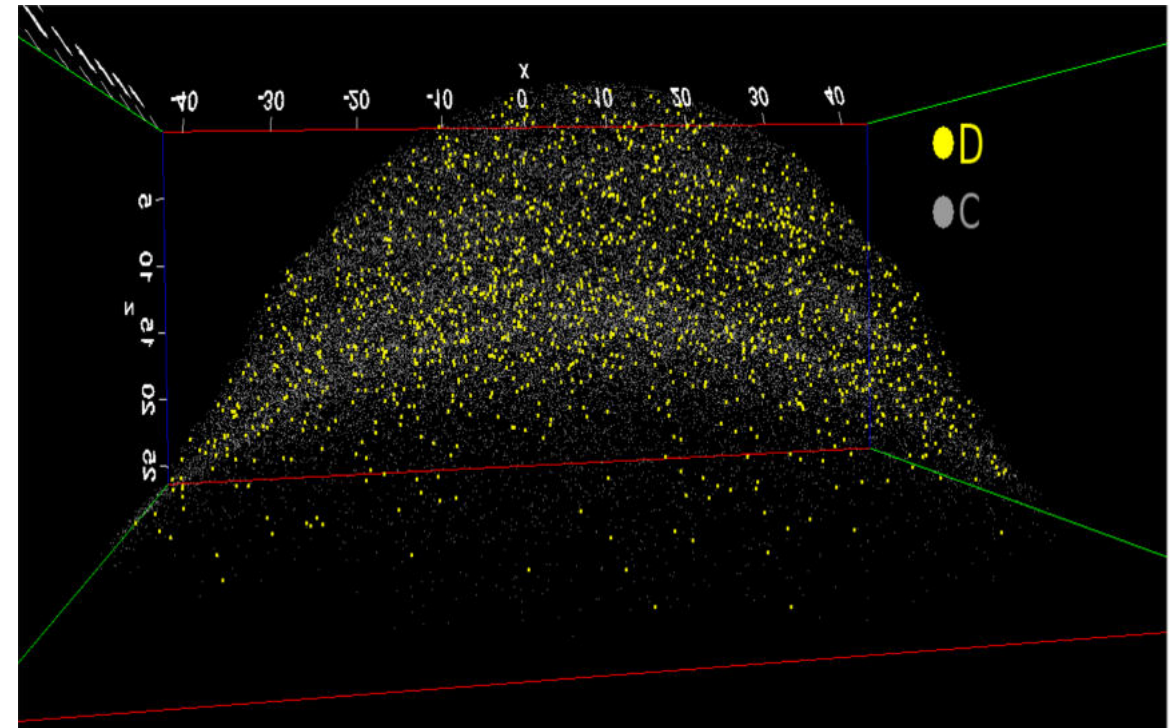


Before D charging (15 hrs aged)



- ❖ Variation in C concentration – spinodal decomposition

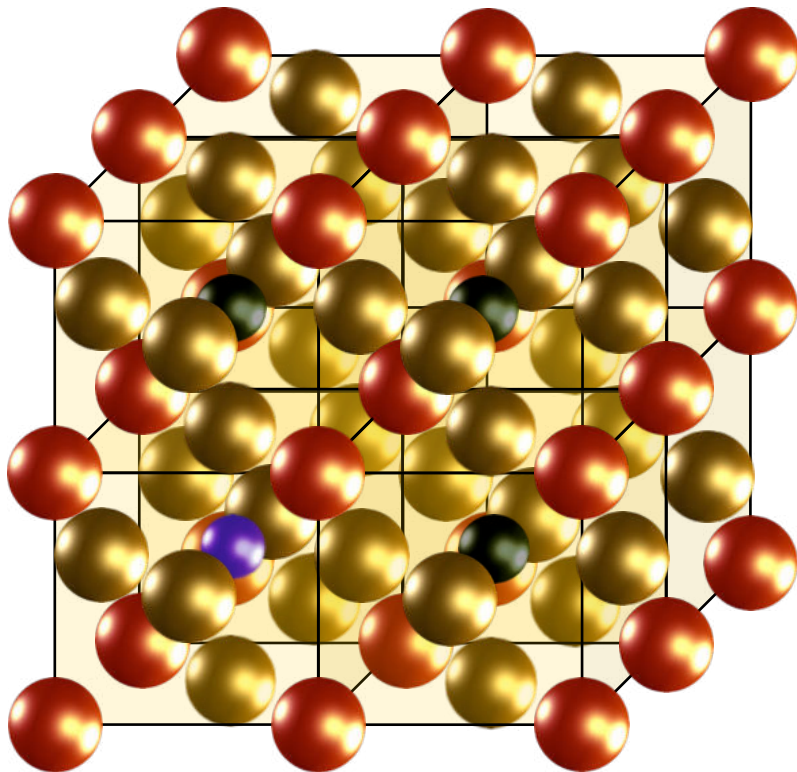
After D charging (15 hrs aged)

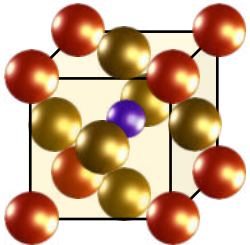


- ❖ No clear D localization to C-rich region visible

Significant amount of Deuterium in the sample – precipitate increases H solubility!

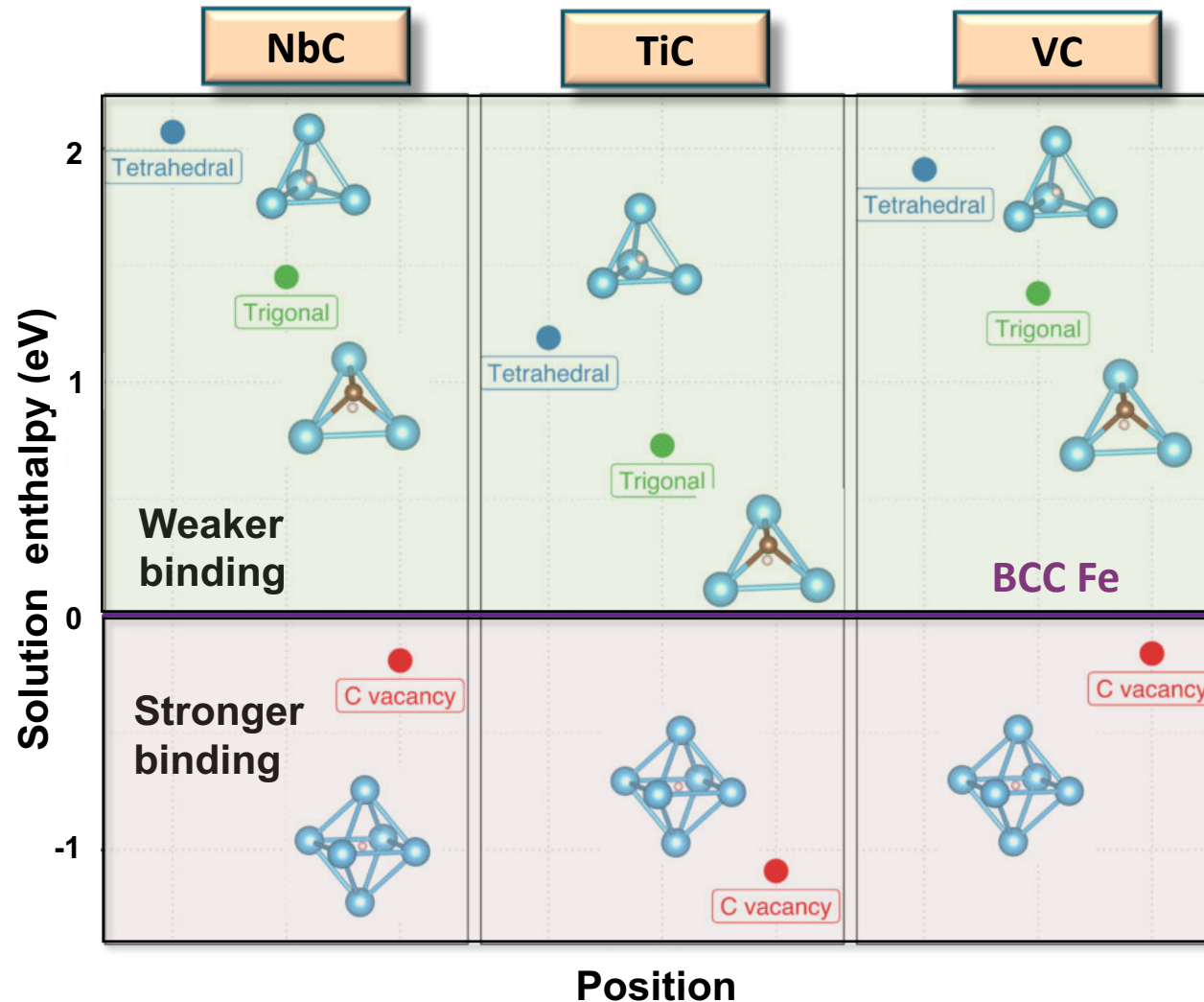
Hydrogen solubility in FeAlC: $\Delta H_H = E[\text{FeAlCH}] - E[\text{FeAlC with one vacancy}] - \mu_H$



H in OS	Solution enthalpy (eV)
	-0.30

C-vacancies in κ carbides trap H!

COMPARATIVE DFT STUDY: H TRAPPING

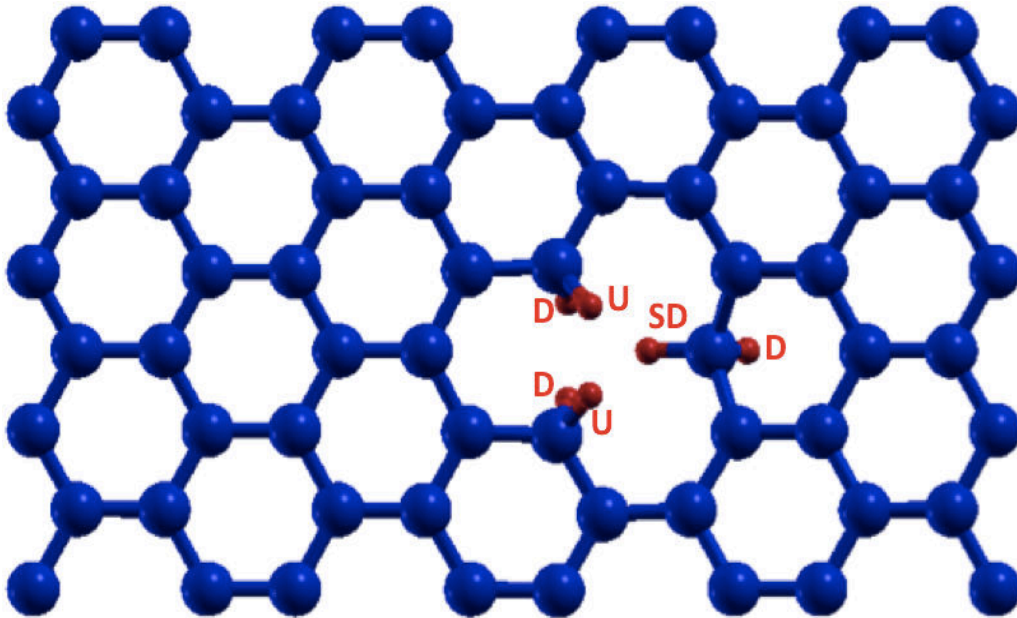


➤ Negative solution enthalpy means H dissolution from Fe into carbide is exothermic

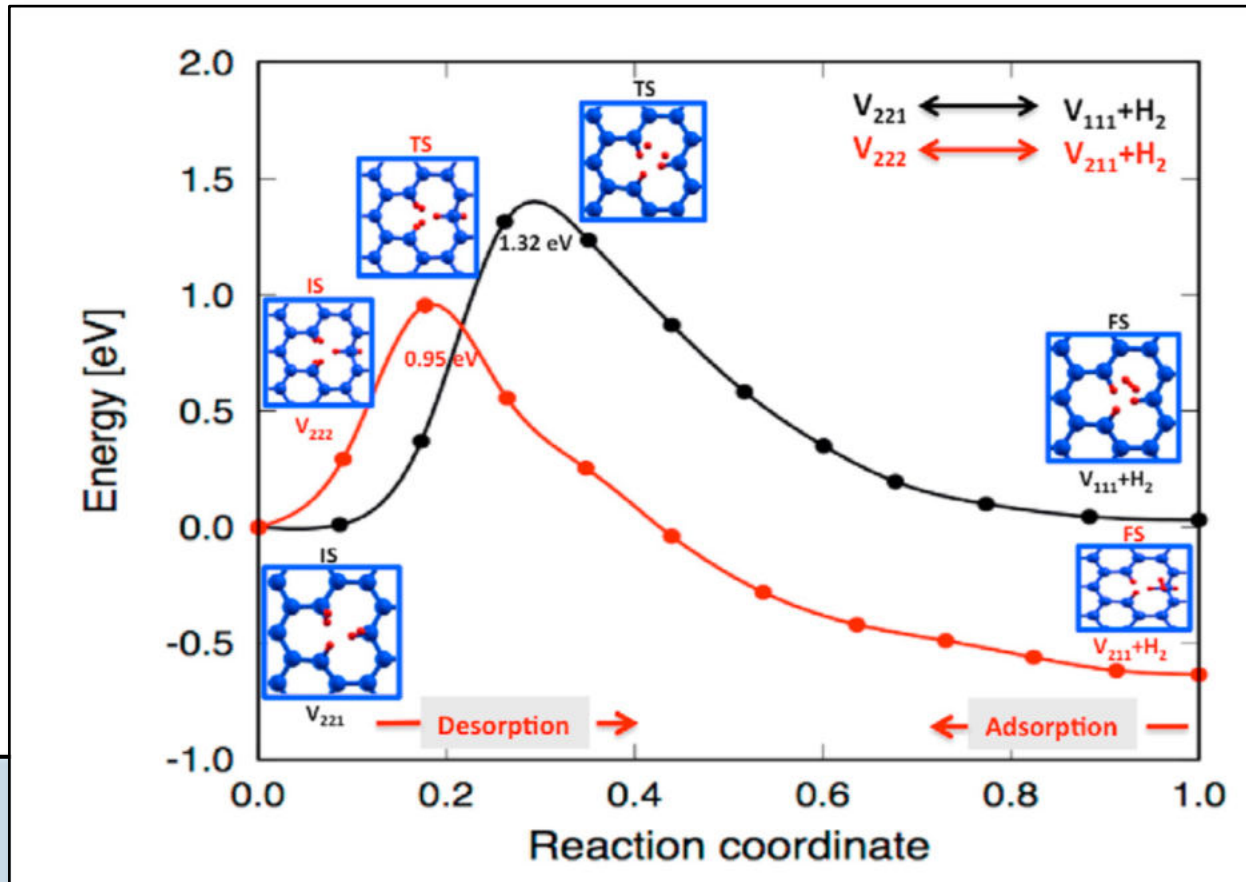
➤ Dissolution in interstitial sites endothermic - trigonal site favoured over tetrahedral site

➤ Strong binding at C vacancy sites - Comparable solution enthalpies for NbC and VC and exceptionally high for TiC

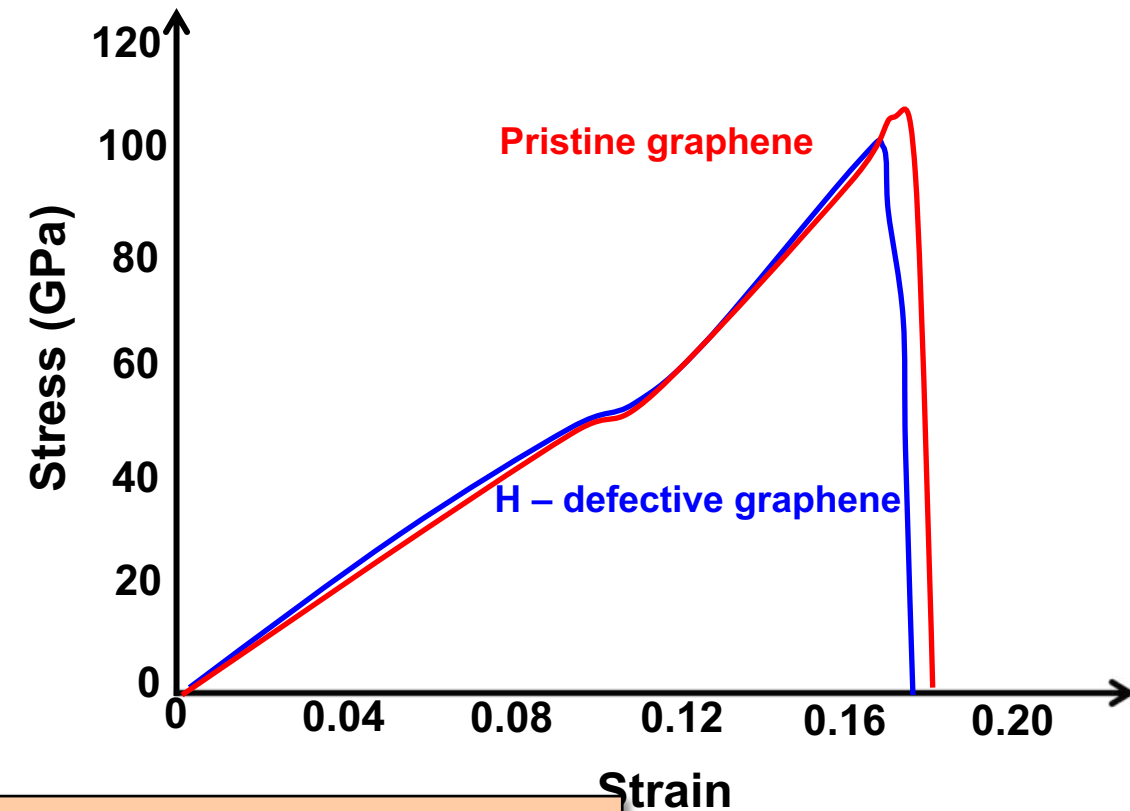
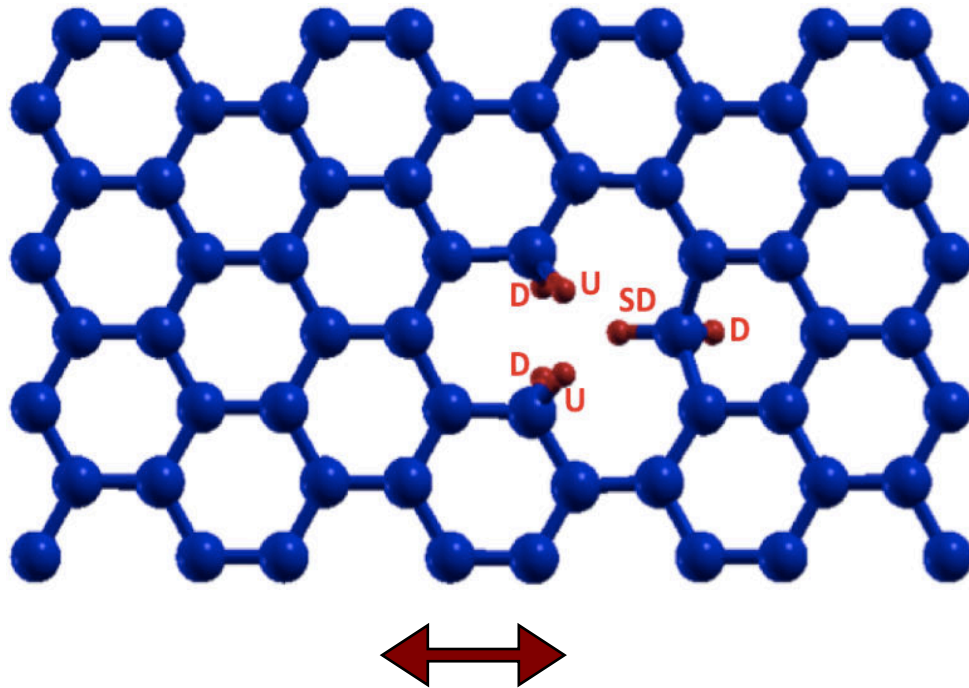
EXAMPLE: GRAPHENE



Energy barriers associated with H decorated defective graphene



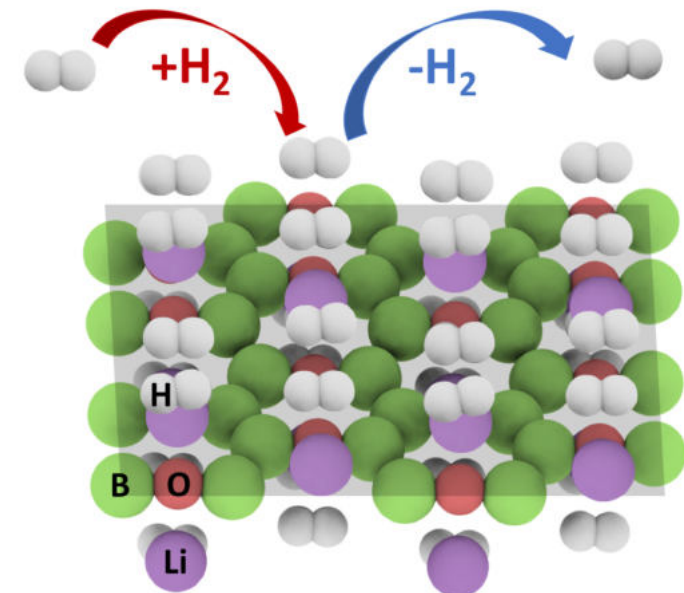
EXAMPLE: GRAPHENE

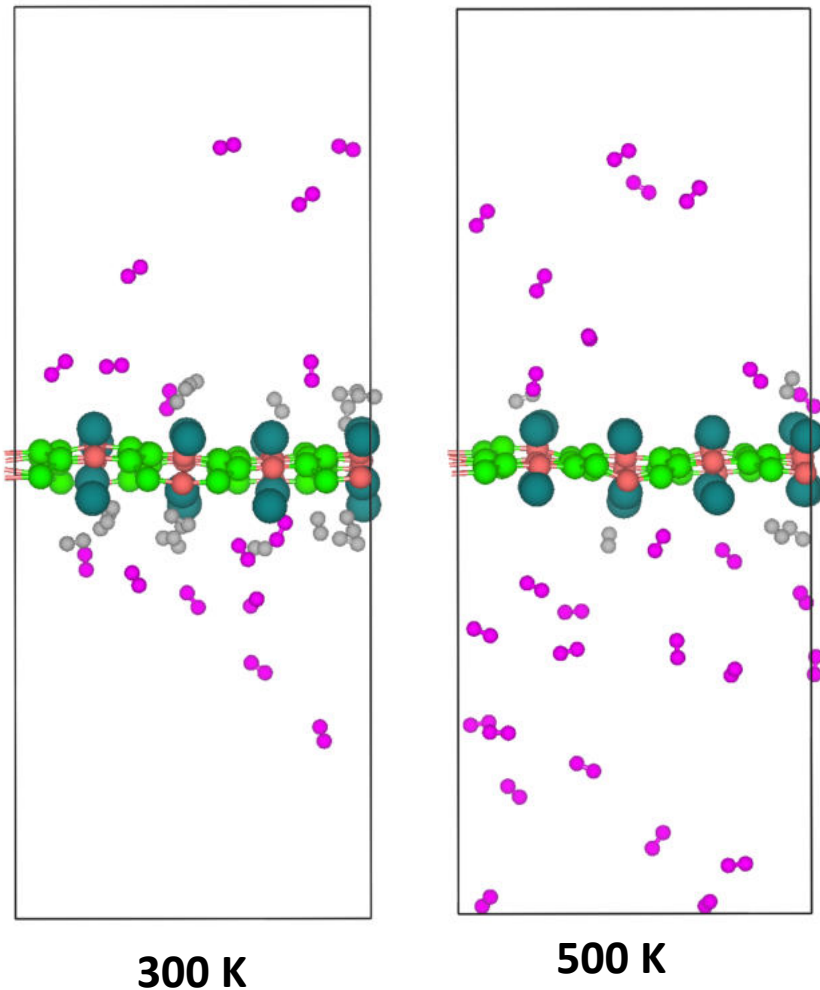


- Not much effect of the presence of H on stress-strain curve
- Slight reduction in the value of the fracture strain and tensile strength

EXAMPLE: BOROPHENE OXIDE WITH LITHIUM

- Li is introduced to enhance interactions with H_2
- Attractive theoretical hydrogen gravimetric density is obtained
- Ab-initio molecular dynamics simulations to test stability and H_2 desorption at finite temperatures





EXAMPLE: BOROPHENE OXIDE WITH LITHIUM

- Each Li binds to two H₂ molecules – average binding energy of -0.24 eV/H₂
- Structural stability tested at temperatures of 200, 300, 400 and 500K
- H₂ bound by van der Waals forces - structure does not undergo irreversible changes
- 53% of H₂ still bound at 300K - 19% bound at 400K

Transition to sustainable sources of energy



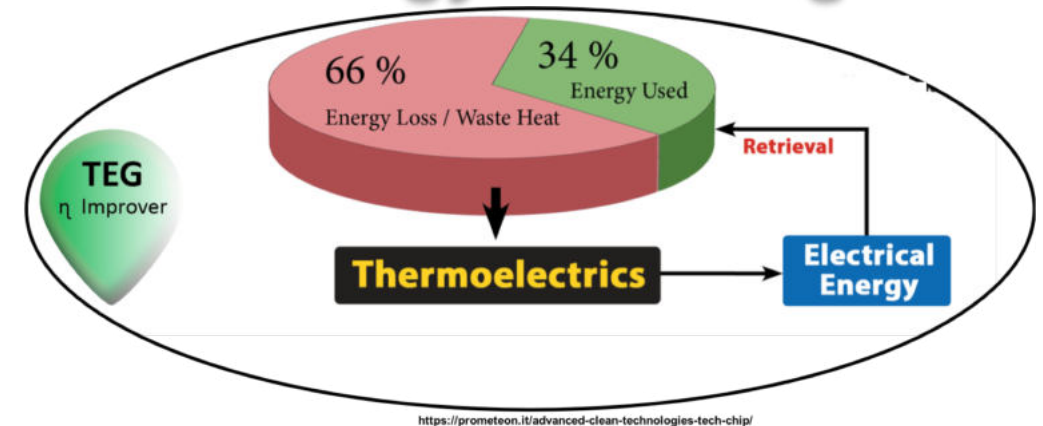
Hydrogen Energy



Storage & Transport

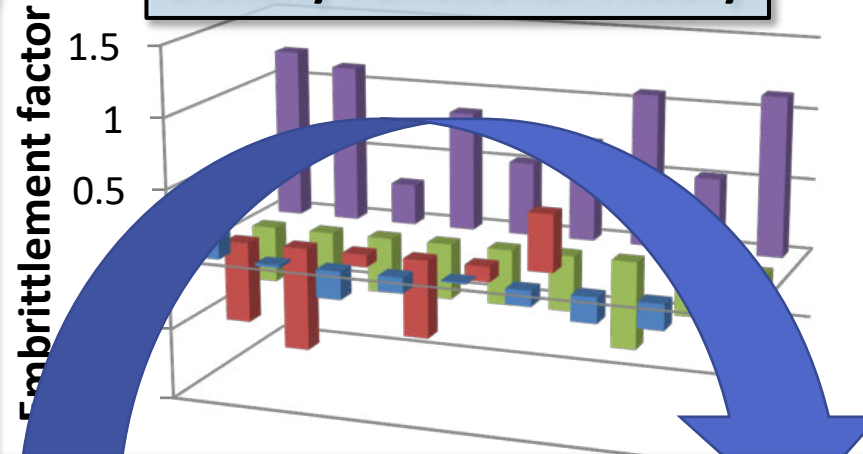


Thermoelectricity – energy & cooling



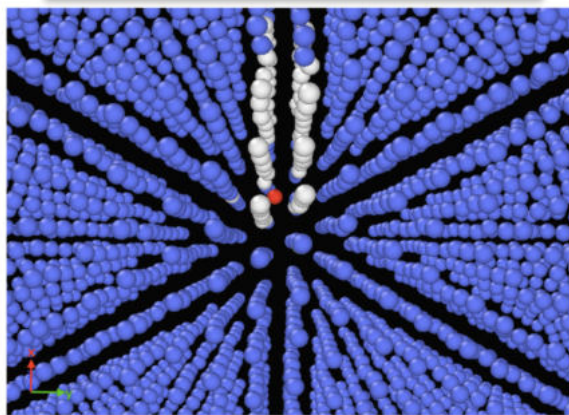
Heat to energy
conversion

Density Functional Theory

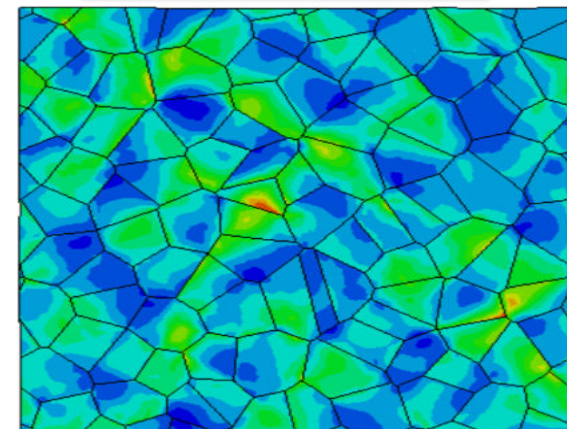


Multi-scale framework to design materials for energy application

Molecular Dynamics



Crystal Plasticity



TEAM

Thank You!

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Dr. Gagus Sunnardianto



Mr. Parsa Habibi



Mr. Tim Boot



Mr. Saurabh Sagar

